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Answer

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Atomic Radius,
Ionization~~

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tronegativity*

Answer
How to Write the
Electron

Configuration
for an Element
in Each Block

*How To Memorize
The Periodic
Table - Easiest
Way Possible*

(Video 1)
Periodic Trends:
Electronegativit

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Study, Ionization

Energy, Atomic

Radius - TUTOR

HOTLINE *Periodic*

Table Explained:

Introduction

Learn the Basics

of the Periodic

Table! Atoms

~~\u0026 the~~

~~Periodic Table~~

~~(updated) Atomic~~

~~Number, Atomic~~

~~Mass, and the~~

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~~Atomic Structure~~

~~| How to Pass~~

~~Chemistry How To~~

~~Memorize The~~

~~Periodic Table~~

~~Through~~

~~Practice!~~

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the Periodic

Table with

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to Pass

Chemistry

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~~Introduction to
the Periodic~~

~~Table~~ **The**

**(truly) Periodic
Table**

Understanding

*Atoms, elements,
and molecules*

Part #1 (9min)

Page 10/49

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Electron Guide

Geometry,

Molecular

Geometry \u0026

Polarity Atoms

and the Periodic

Table *Electron*

Configuration -

Basic

introduction

Intro to

Chemistry, Basic

Concepts -

Periodic Table,

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Elements, Metric
System \u0026amp;

Unit Conversion

Periodic Table

of Elements

Explained -

Metals,

Nonmetals,

Valence

Electrons,

Charges FORM 2

CHEMISTRY THE

STRUCTURE OF AN

ATOM \u0026amp; THE

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PERIODIC TABLE

~~Chemistry~~

~~Atomic Structure~~

~~— EXPLAINED! —~~

~~Modern Periodic~~

~~Table Chemistry~~

Tutorial: Atomic

Structure meets

the Periodic

Table Atoms

~~Periodic Table~~

~~Study Guide~~

Unit 4 Atoms and

the Periodic

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Table Study

Guide 1. The
smallest

"uncuttable"
particle of an
element is the
_____. 2. Atoms
of the same
element have
exactly the same
_____. Atoms of
different
elements have
different _____.

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Answer

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the Periodic
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Guide~~

PPT_ Atoms & the
Periodic Table
Notes_ Atoms &
the Periodic
Table Nathan's
PPT - Electron
Configuration
Atomic Structure

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& the Periodic
Table; PhET
Simulation:

Build an Atom;
Exploring the
Periodic Table;
Extra Practice -
Using the
Periodic Table;
ChemThink:

Atomic
Structure;
Video: How does
an Atom look

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like? Ions & Ion
Notation (Atom
with a Charge)

~~3. Atoms & the
Periodic Table~~

~~Mrs. Sumerix~~

The periodic
table helps
predict some
properties of
the elements
compared to each
other. Atom size

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decreases as you
move from left
to right across
the table and
increases as you
move down a
column. The
energy required
to remove an
electron from an
atom increases
as you move from
left to right
and decreases as

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you move down a
column.

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Introduction &
History~~

Honors Study
Guide- Atoms and
the Periodic
Table of
Elements 1.

Define Element:
An element is a

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chemical Study Guide

substance that
is in the form
of an atom 2.

Define Atom: An
atom is small
sub-atomic
particles held
together by
fundamental
forces 3.

Describe changes
in the atomic
model over time

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and why those
changes were
necessitated by
experimental
evidence.

Explain how each
discovery lead
to the modern

...

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riodic_Table_Stu
dy_Guide.docx
Honors...~~

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Study Guide -

Atoms & the
Periodic Table

Name _____ The
Scientists A.

Thomson's study
of cathode rays
revealed that
atoms contained
negatively
charged
particles called
_____ . B.

Rutherford's

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gold-foil
Study Guide

experiment

Answer

revealed that an
atom's positive
charge is
concentrated in
the atom's

~~Study Guide~~

~~Atoms & the~~

~~Periodic Table~~

Name

The position of
elements on the

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periodic table
is directly
related to their
electron shells.
Describe the
relationship
between group
number and
number of
valence
electrons. the
group number is
how many valence
electrons of

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that element are
in each atom.

Describe the
role of a
coefficient in a
chemical
equation.

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Study & the
Answer

Periodic Table

Subpages (7):

Lesson 1 Lesson

1: Parts of the

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Atom Lesson 3:

Ions & Isotopes

Lesson 4:

Electron Locatio

n/Configuration

Lesson...

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~~Unit 4: Atoms &
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~~Physical Science~~

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table KS3
Chemistry ...~~

The study guide
for this unit is
provided here:

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Atoms and
Elements 4)

Daily Warm Ups -
Daily warm ups,
"do it nows" or
"bell

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serve the
purpose of
engaging
students in the
daily learning,
focusing
students for
scientific study
and to practice
skills or learn
concepts.

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~~Atoms and~~

~~Elements~~

~~Better Lesson~~

Mass number. sum
of the number of
protons and
neutrons; number
of protons +
number of
neutrons = mass
number, number
of neutrons =
mass number -
number of

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Study Guide
Answer

protons (atomic number), number of protons = number of electrons IF positive and negative charges cancel, the atom Charge = 0 (Stable state, non ionic state)
Isotopes.

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Table; Atoms
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Electron Guide

affinity
Answer
increases from
left to right
across the
periodic table.
This is caused
by the decrease
in atomic
radius. As we
already
explained,
moving from left
to right across

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a period, atoms become smaller and smaller as the atomic number increases. This causes the electron to move closer to the nucleus, thus increasing the electron affinity.

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Trends Help |
The Periodic
Table Study ...~~

Atoms and the
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notes and use
your Periodic
tables This quiz
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enter your Quia

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username and
password.

~~Quia Study
Guide for Atoms
and the Periodic
Table~~

Each row in the
periodic table
is called a
period and the
period number
tells how many
energy levels

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atoms in that
period have. As
a result, as you
go down a group
in the periodic
table the
sizes...

~~Describe atomic
size. How does
it relate to the
periodic ...~~

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