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nm), E_0 6.13 eV), and n (10), and asks that we determine explicit expressions for attractive and repulsive energies of Equations 2.8 and 2.9. In essence, it is necessary to compute the values of A and

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B in these equations.

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solved using two steps: (1) calculate the total number of lattice sites in silver, N_{Ag} , using Equation 4.2; and (2) multiply this number by fraction of lattice that are vacant, 2×10^{-6} . The parameter N_{Ag} is related to

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the density, (), Avogadro's number (N_A), and the atomic weight ($A_{Ag} = 107.87 \text{ g/mol}$, from inside the

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Equation 4.2; and (2)
multiply this number by
fraction of lattice that are
vacant, 2

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Solution The average atomic weight of silicon (A) is computed by adding fraction-of-occurrence/atomic weight Si products for the three isotopes—i.e., using Equation 2.2. (Remember: fraction of occurrence is

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equal to the percent of occurrence divided by 100.)

Thus. $A_{Si} = \frac{f_{28} A_{28} + f_{29} A_{29} + f_{30} A_{Si}}{Si_{Si}}$

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L L + ? ? 0 20 40 60 80 100
wt% Ni 1000 1100 1200 1300
1400 1500 1600 T (°C) L

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