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18 States of Matter

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You are already familiar with the three common states of matter: solid,

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liquid, and gas. Solid objects litter the room around you. For example, you can easily recognize the shape of your desk; you know that your backpack cannot hold seven textbooks. You encounter liquids throughout the day as you

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Chapter 13 States of
Matter 137 SECTION

13.1 THE NATURE OF
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theory and describes
how it applies to gases.

It defines gas pressure
and explains how
temperature is related to
the kinetic energy of the
particles of a substance.

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Kinetic Theory and a
Model for Gases (pages
385–386) 1.

Name Date Class

STATES OF MATTER

13

There are three states of matter that we will learn about in this chapter. (If you want to learn about more states of matter, I can refer you to somebody.) Those three

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Chapter 13

states are solid, liquid, and gas. ? These three states are quite different. The main difference is in their particles.

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Chapter 13 States Of Matter Chapter 13 States of Matter137 SECTION 13.1 THE NATURE OF GASES

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(pages 385–389) This section introduces the kinetic theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance.

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SECTION 13.1 THE NATURE OF GASES

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energy of the particles
of a substance.

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Chapter 13: States of
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Kinetic Molecular

Theory. Explains the
properties of gases in

terms of the energy,

size, and motion of their
particles. Elastic

Collision. Describes a

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collision in which kinetic energy may be transferred between colliding particles but the total kinetic energy of the two particles remains the same.

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Matter - 13.1 The Matter

Nature of Gases - 13.1

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Publisher: Prentice Hall

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Chapter 13

Edition) Chapter 13 -

States of Matter ...

all matter consists of tiny particles that are constantly in motion

What are the three assumptions of the kinetic theory as it applies to gases? -The particles in a gas are considered to be small, hard spheres with an insignificant volume.

-The motion of the

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Chapter 13

particles in a gas are rapid, constant, and random.

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The Sustainable Development Goals are a call for action by all countries – poor, rich and middle-income – to promote prosperity while protecting the

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planet. They recognize
that ending poverty ...

United Nations

Sustainable

*Development – 17 Goals
to ...*

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You; 13.1 The Nature of
Gases - Sample Problem
13.1; 13.1 The Nature of
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Check; 13.2 The Nature of Liquids - Chemistry & You; 13.2 The Nature of Liquids - 13.2 Lesson Check; 13.3 The Nature of Solids - Chemistry & You; 13.3 The Nature of Solids - 13.3 Lesson Check; 13.4 Changes of State - Chemistry & You

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of Matter 1 Chapter 13
States of Matter 2
Kinetic Theory as
Applied to Gases
Fundamental
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gases. The particles in a gas are considered to be small, hard spheres with an insignificant volume.

Between particles in a gas there is empty space. No attractive or repulsive forces exist between the particles. 3

Chapter 13 States Of Matter Worksheet

The attacks the USPS continues to face are not

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just attacks on the postal service but attacks on Black lives. To defund the USPS would be to deny future generations this opportunity and dishonor the legacy of Black postal workers. Now, we're taking this matter into our own hands by writing and sending #BlackLoveLetters through USPS...

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