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This includes
breaking and/or
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between elements.

Let's look at a
chemical reaction
using Lewis Dot

Diagrams to
visualize it. The
synthesis reaction
above shows an
interaction

between
magnesium and
oxygen. The bonds
that are broken are
red, and the bonds

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formed are green.

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IDEA: Chemical

reactions release

or absorb energy.

6. The of the

reactants and

products

determines

whether energy will

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be released or absorbed during a chemical reaction.

7. Before a chemical reaction can start, must be absorbed by the reactants. The amount that must be absorbed to start the reaction is called the.. 8.

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2.4 Chemical
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Aaron Reabe PART
1: Count the atoms
in each of the
following chemical
formulas. 2) 2NaCl
2) 2MgCO_3 3) 3OH_2
2) 3OH 4) 4CaCl_2
Cl 4 10 39 2 24
PART 2: Each of the

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following
statements is false.
Underline the ONE
word that is
incorrect. A. If a
student adds 5
grams of baking
soda to 50 grams
of vinegar, the
product will equal
50 grams. B.

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A chemical change or chemical reaction is a process in which one or more pure substances are converted into one or more different pure substances. 9. A chemical equation is a shorthand description of a chemical reaction.

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Cornelius about
Peter. 1. (Act
10:1-2) Cornelius, a
Gentile who served
God. There was a
certain man in
Caesarea called
Cornelius, a
centurion of what
was called the
Italian Regiment, a
devout man and

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and Reactions
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MIXED REVIEW
ANSWERS

Answer the
following questions
in the space
provided. 1. b A
balanced chemical
equation

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represents all the following except

- (a) experimentally established facts.
- (b) the mechanism by which reactants combine to form products.
- (c) identities of reactants and products in a chemical reaction.

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