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Science Revision

Chemistry \ "The Three
States of Matter\ "

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Science Guy Phases of

Matter Homogeneous

and Heterogeneous

Mixtures Examples,

Classification of Matter,

Chemistry Chemistry

Matter and Change Part

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Matter And

Lifewave Webinar 1.1

Pure Substances and
Mixtures, Elements

Compounds,

Classification of Matter,

Chemistry Examples,

3.2.1/3.2.2 Describe the
differences between

elements, compounds

and mixtures.

~~States of
Matter : Solid Liquid~~

~~Gas 10 Amazing~~

~~Experiments with Water~~

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Changes in the State of
Matter - Science

Experiments - Solid to
Liquid, Liquid to Gas,

Gas to Liquid Changes
of States of Matter

Conservation of Mass in
Chemical Reactions The

States of Matter -

Changes of State - Solid
Liquid Gas Plasma |

Physical Science | Gen
Chemistry States of

Matter *Understanding*

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*Atoms, elements, and
molecules Part #1*

(9min) Study Jams -

Properties of Matter

States of Matter and

Changes of State -

Science for Kids

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Dr. Binocs Show | Best

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Pure Substances and

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(Classification of

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Great Picnic Mix Up:

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#19.1 Phase Changes

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Quanta, Matter &

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Approach to Physical
Chemistry-Charles

Trapp 2009-09-10 This
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Quanta, Matter &
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worked solutions to the
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very helpful. Thank you
so much crazy for study
for your amazing
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**Chemistry: The
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5th ...

A change in physical form of a substance or physical properties of a substance like color, melting point, state, size, shape, and arrangement of molecules is known as physical change. A

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change in chemical properties of a substance like composition, reactivity with water, corrosiveness, and pH is known as chemical change. In chemical change, a new substance will be formed.

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7th ...**

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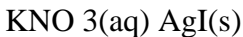
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CHAPTER 9

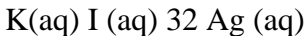
SOLUTIONS

MANUAL 35. Aqueous
solutions of potassium
iodide and silver nitrate
are mixed, forming the
precipitate silver iodide.

chemical equation:



complete ionic equation:



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$\text{NO}_3^-(\text{aq}) + \text{K}^+(\text{aq}) + \text{NO}_3^-(\text{aq}) + \text{AgI}(\text{s})$

$3^-(\text{aq}) + \text{AgI}(\text{s})$

Solutions

Chemical

Reactions

Solution. Physical:

boiling and melting are physical changes. When water boils, no bonds are broken or formed.

The change could be written: $\left(\text{H}_2\text{O} \right)$

$\left(1 \right)$

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$\rightarrow \text{H}_2\text{O}$
(g)

Chemical: The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

3.6: Changes in Matter: Physical and Chemical Changes ...

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CHAPTER 12

SOLUTIONS

MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in

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a liquid gain enough

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