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Answer Key 2.1 The Nature of Matter

Lesson Summary Atoms The atom is the basic unit of matter, made up of three subatomic particles. Protons have a positive charge and neutrons carry no charge. Strong forces bind protons and neutrons together in the nucleus.

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how and why substances combine or separate to form other substances, and how According to the ACS, chemistry is the study of matter, defined as anything that has mass and takes up space, and the changes that matter can undergo when it is...

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Chapter 8 • The partial pressure of a gas depends on the number of moles, size of the container, and temperature and is independent of the type of gas.

- At a given temperature and

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pressure, the partial pressure of 1 mol
of any gas is the same.

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Type of mixture NOT uniform throughout (clay and water mixture - clay gathers at bottom) Pure substance. Has fixed composition and differs from a mixture in that: 1- every sample has exact same characteristic physical and chemical properties used

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to identify substance. 2- every sample
of a pure substance has exact same
composition with the same makeup
(i.e.: pure water is always 11.2% H and
88.8% oxygen by mass.

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assistance for doctors as a really
should for independently practicing
physicians and little and medium
stage hospitals. The leading advantage
of choosing a call answering supplier

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explains the properties of the different states of matter.

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In the space at the left, write true if the statement is true; if the statement is false, change the italicized word or phrase to make it true 8 To determine the

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first ionization energies generally
increase as you move left-to-right

across a period. The increased nuclear
charge of each successive element

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produces an increased hold on the valence electrons. 10. The first ionization energies generally decrease as

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