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Chang, Raymond : Williams College. Raymond Chang was born in Hong Kong and grew up in Shanghai and Hong Kong, China. He received his B.Sc. degree in chemistry from London University, England and his Ph.D. in physical chemistry from Yale University.

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This book is ideal for use in a one-semester introductory course in

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physical chemistry for students of life sciences. The author's aim is to emphasize the understanding of physical concepts rather than focus on precise mathematical development or on actual experimental details. Subsequently, only basic skills of differential and integral calculus are required for understanding the equations. The end-of-chapter problems have both physiochemical and biological applications.

Hailed by advance reviewers as "a kinder, gentler P. Chem. text," this book meets the needs of an introductory course on physical chemistry, and is an ideal choice for courses geared toward pre-medical and life sciences students. Physical Chemistry for the Chemical and Biological Sciences offers a wealth of applications to biological problems, numerous worked examples and around 1000 chapter-end problems.

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The fifth edition of General Chemistry continues the tradition of presenting only the material that is essential for a one-year general chemistry course. It strikes a balance between theory and application by incorporating real-world examples; helping students visualize the three-dimensional atomic and molecular structures that are the basis of chemical activity; and developing problem-solving and critical thinking skills. Although the fifth edition incorporates many new features, such as macro to micro artwork, six new animations correlated to the text, and the addition of new hand-sketched worked examples, General Chemistry is still 200 to 300 pages shorter and much less expensive than other two-semester textbooks. Dr. Chang's concise-but-thorough approach will appeal to efficiency-minded instructors

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and value-conscious students.

Chang's best-selling general chemistry textbook takes a traditional approach and is often considered a student and teacher favorite. The book features a straightforward, clear writing style and proven problem-solving strategies. It continues the tradition of providing a firm foundation in chemical concepts and principles while presenting a broad range of topics in a clear, concise manner. The tradition of "Chemistry" has a new addition with co-author, Kenneth Goldsby from Florida State University, adding variations to the 12th edition. The organization of the chapter order has changed with nuclear chemistry moving up in the chapter order.

The third edition of General Chemistry: The Essential Concepts

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continues the tradition of presenting only the material that is essential for a one-year general chemistry course. As before, the text includes all the core topics that are necessary for a solid foundation in general chemistry without sacrificing depth, clarity, or rigor. The third edition features new Spartan molecular models, animations correlated to the text, the addition of molecular orbital theory and more. General Chemistry: The Essential Concepts is 200 to 300 pages shorter than traditional two-semester textbooks and is much less expensive. Dr. Chang's concise-but-thorough approach will appeal to efficiency-minded instructors and will please value-conscious students.

The importance of metals in biology, the environment and medicine

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has become increasingly evident over the last twenty five years. The study of the multiple roles of metal ions in biological systems, the rapidly expanding interface between inorganic chemistry and biology constitutes the subject called Biological Inorganic Chemistry. The present text, written by a biochemist, with a long career experience in the field (particularly iron and copper) presents an introduction to this exciting and dynamic field. The book begins with introductory chapters, which together constitute an overview of the concepts, both chemical and biological, which are required to equip the reader for the detailed analysis which follows. Pathways of metal assimilation, storage and transport, as well as metal homeostasis are dealt with next. Thereafter, individual chapters discuss the roles of sodium and potassium, magnesium, calcium, zinc, iron, copper, nickel and cobalt, manganese, and finally molybdenum, vanadium, tungsten and

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chromium. The final three chapters provide a tantalising view of the roles of metals in brain function, biomineralization and a brief illustration of their importance in both medicine and the environment. Relaxed and agreeable writing style. The reader will not only find the book easy to read, the fascinating anecdotes and footnotes will give him pegs to hang important ideas on. Written by a biochemist. Will enable the reader to more readily grasp the biological and clinical relevance of the subject. Many colour illustrations. Enables easier visualization of molecular mechanisms Written by a single author. Ensures homogeneity of style and effective cross referencing between chapters

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