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Activity A

Practice balancing chemical equations by

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changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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directions towards clinic and answering a variety of other routine issues from patients. This function of handling calls can easily be delegated into a team experienced in taking and answering calls for the doctors.

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This process is summarized by a chemical

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Answers. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$. 1. How many atoms are in a hydrogen molecule (H_2)? 2 2. How many atoms are in an oxygen molecule (O_2)? 2 3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)? H:

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Answers: In general, what does a subscript
(such as the “ 2 ” in H

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formulas using the Gizmo. AgCl_3Cu_2 . Ag:
_ 1 _ Cl: _ 3 _ . Cu: _ 2 _ . Ba

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 12 P: 1 ...

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Balancing Chemical Equations Worksheet

1 - ANSWERS 1. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

2. $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$ 3. $\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$

4. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ 5. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

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If you also get perplexed in balancing chemical equations, follow the tips for correct balancing chemical equations worksheet answers. Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value of coefficient in front of the element or compound, and not

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