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Answer the following questions in the
space provided. 1. Given the following

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Equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO$

$2(g) + 2H_2O(g)$ 4 a. What is the value
of the coefficient x in this equation?

40.07 g/mol b. What is the molar mass
of C_3H_4 ? 2 mol O_2 :1 mol H_2O c.

What is the mole ratio ...

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Stoichiometry Answers Mix & match

(both simple stoichiometry and limiting reagent problems) Units to use (select at least one): Grams Moles Particles

(e.g. atoms/molecules/formula units)

Chemical formulas or names:

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Formulas only Names only Mix &
match formulas &

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SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1.

_____ The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol

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_____ The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage ...

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Chapter 9 Stoichiometry. composition stoichiometry deals with the mass relationships of elements in compounds. reaction stoichiometry involves the mass relationships between reactants and products in a chemical reaction.

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HOMEWORK 9-2 (pp. 280-282)
VOCABULARY Complete each
sentence. 1. ... the provided a 1

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