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1. Acetylsalicylic acid, the common mild pain reliever, is a derivative of the natural

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Chapter 7: Chemical Equilibrium 7.5-2

Statement: The value of  $Q$  is 0.867.  $Q$  is approximately equal to  $K$ , so the reaction is at equilibrium. 3. (a) The equilibrium constant equation for the reaction represented by  $! 2 Ag 2O(s) \rightleftharpoons 4 Ag(s) + O 2(g)$  is  $K=[O 2 (g)]$ , because the other substances are solids.

~~Section 7.5: Quantitative Changes in  
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By Paulo Coelho cuso4s 5 h2ol h2g i2g 2

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The video below refers to Dynamic Equilibrium (under the section Equilibrium Systems) 7.1 p. 420 in the Nelson Textbook The video below is related to 7.2 Equilibrium Law and the Equilibrium Constant (p.429 in the Textbook)

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f 2 3.5 The conservation of kinetic energy equation can be simplified by multiplying

both sides of the equation by 2 and noting

that  $v_i^2 = 0$  m/s. Section 5.3: Collisions

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12 - Chapter 7 Quiz. True/False. Indicate

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