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Solution From Coulomb's law of electrostatic attraction we know that the PE between two charges Q_1 and Q_2 separated by a distance r is given by $PE = \frac{Q_1 Q_2}{4\pi\epsilon_0 r}$ First we consider the interaction between Na^+ ion and

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6Cl⁻ ions at distance r Applying Coulomb's law we have Q_1Q_2
($6e$)($+e$) $6e$ PE₁ = = = $4??$ o r $4??$ o r $4??$ o r Similarly, we now consider 12 Na⁺ ions as second nearest neighbors at a distance r Q_1Q_2 ($+12e$)($+e$) $12e$ = PE₂ = = $4??$ o r $4??$ o r $4??$ o r and Cl ...

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2.1 Electrical conduction Na is a monovalent metal (BCC) with a density of 0.9712 g cm^{-3} . Its atomic mass is 22.99 g mol^{-1} .

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