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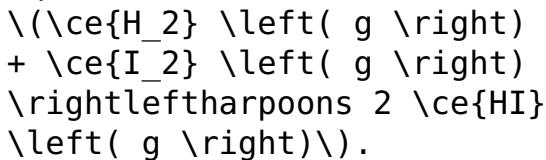
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**Equilibrium Chemistry** is the state of a system in which the rate of the forward reaction is equal to the rate of the reverse reaction. Figure

\(\backslash\PageIndex{1}\backslash\):

Equilibrium in reaction:



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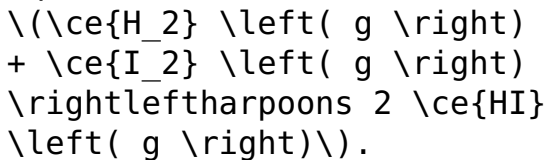
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## Equilibrium Chemistry

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Equilibrium in reaction:



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Equilibrium Once this has  
happened for awhile, there  
is a build up of  $\text{N}_2$

molecules in the same flask

Once in awhile, two  $\text{N}_2$

molecules will collide with  
each other and join to form

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reactions do not go to  
completion but instead  
attain a state of chemical  
equilibrium Chemical  
equilibrium: A state in  
which the rates of the  
forward and reverse  
reactions are equal and the

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in fact, we will spend much of the rest of the course discussing aspects of this concept. solution manual physical and chemical equilibrium for solution manual chemistry : an everyday approach to chemical investigation (sally solomon, susan rutkowsky & charles

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A solution equilibrium occurs when a solid substance is in a saturated solution. At this point, the rate of dissolution is equal to the rate of recrystallization. Although

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Equilibrium Chemistry  
these are all different types of transformations, most of the rules regarding equilibrium apply to any situation in which a process occurs reversibly.

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to completion but instead  
attain a state of chemical  
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equilibrium: A state in  
which the rates of the

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